

## Spatial and temporal variability in summertime dissolved carbon dioxide and methane in temperate ponds and shallow lakes

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### Abstract

Small waterbodies have potentially high greenhouse gas emissions relative to their small footprint on the landscape, although there is high uncertainty in model estimates. Scaling their carbon dioxide (CO<sub>2</sub>) and methane (CH<sub>4</sub>) exchange with the atmosphere remains challenging due to an incomplete understanding and characterization of spatial and temporal variability in CO<sub>2</sub> and CH<sub>4</sub>. Here, we measured partial pressures of CO<sub>2</sub> (*p*CO<sub>2</sub>) and CH<sub>4</sub> (*p*CH<sub>4</sub>) across 30 ponds and shallow lakes during summer in temperate regions of Europe and North America. We sampled each waterbody in three locations at three times during the growing season, and tested which physical, chemical, and biological characteristics related to the means and variability of *p*CO<sub>2</sub> and *p*CH<sub>4</sub> in space and time. Summer means of *p*CO<sub>2</sub> and *p*CH<sub>4</sub> were inversely related to waterbody size and positively related to floating vegetative cover; *p*CO<sub>2</sub> was also positively related to dissolved phosphorus. Temporal variability in partial pressure in both gases was greater than spatial variability. Although sampling on a single date was likely to misestimate mean seasonal *p*CO<sub>2</sub> by up to 26%, mean seasonal *p*CH<sub>4</sub> could be misestimated by up to 64.5%. Shallower systems displayed the most temporal variability in *p*CH<sub>4</sub> and waterbodies with more vegetation cover had lower temporal

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Additional Supporting Information may be found in the online version of this article.

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variability. Inland waters remain one of the most uncertain components of the global carbon budget; understanding spatial and temporal variability will ultimately help us to constrain our estimates and inform research priorities.

Lentic waterbodies play a major role in global carbon dioxide (CO<sub>2</sub>) and methane (CH<sub>4</sub>) cycling (Tranvik et al. 2009; Raymond et al. 2013; Rosentreter et al. 2021a). The smallest of these systems (i.e., ponds) have a particularly outsized influence on global and regional CO<sub>2</sub> and CH<sub>4</sub> budgets relative to larger waterbodies due both to high emissions rates and their ubiquity (Holgerson and Raymond 2016; Ollivier et al. 2019). Despite consensus regarding their importance, global estimates of CO<sub>2</sub> and CH<sub>4</sub> emissions from small aquatic systems are among the most uncertain in global budgets (Canadell et al. 2021) and remain highly variable for several reasons (Raymond et al. 2013; Holgerson and Raymond 2016; Rosentreter et al. 2021a). First, the exact number of ponds and shallow lakes remains unclear due to limitations in mapping ability (Messenger et al. 2016), but there are likely billions of these systems globally (Downing 2010). Second, each system differs in physical, chemical, and biological properties that affect rates of CO<sub>2</sub> and CH<sub>4</sub> exchange with the atmosphere (Laurion et al. 2010; Holgerson and Raymond 2016; Grinham et al. 2018). Third, there is unknown—and often unaccounted for—spatial and temporal variability in dissolved CO<sub>2</sub> and CH<sub>4</sub> concentrations in the surface waters of ponds and shallow lakes.

Considerable progress has been made in understanding the importance of, and controls on, lentic CO<sub>2</sub> and CH<sub>4</sub> cycling and exchange with the atmosphere across space and time (Schilder et al. 2013; Vachon and Prairie 2013; Rudberg et al. 2021). However, the importance of spatial and temporal variability in the smallest of these systems are not as well constrained, and it is even less clear what physical, biological, and chemical properties might be useful for predicting the most variable systems. One of the primary reasons for this lack of understanding is inconsistent sampling protocols, specifically regarding the intensity of spatial and temporal replication. Testing if variables that can predict dissolved CO<sub>2</sub> and CH<sub>4</sub> concentrations can also predict whether a waterbody will have high or low spatial and temporal variability of CO<sub>2</sub> and CH<sub>4</sub> concentrations is an important step in improving our understanding of CO<sub>2</sub> and CH<sub>4</sub> dynamics in small lentic systems and can inform sampling schemes. Determination of the magnitude of error associated with limited sampling will demonstrate whether it is necessary to sample waterbodies across space and time in order to reduce uncertainty in estimating diffusive CO<sub>2</sub> and CH<sub>4</sub> exchange with the atmosphere and inform methods to improve global upscaling efforts (Wik et al. 2016; Natchimuthu et al. 2017; Loken et al. 2019).

Dissolved CO<sub>2</sub> and CH<sub>4</sub> concentrations vary spatially in larger lentic systems (i.e., lakes and reservoirs; Pacheco et al. 2015; Colas et al. 2020; Praetzel et al. 2021). For example, spatial variation in *p*CO<sub>2</sub> was linked with indicators of planktonic primary

production (i.e., dissolved O<sub>2</sub> concentration, pH) while spatial variation in *p*CH<sub>4</sub> was better described by depth and pH in large (> 12 km<sup>2</sup>) constructed Brazilian reservoirs (Paranaíba et al. 2018). Littoral areas of lakes typically have higher CH<sub>4</sub> concentrations than the pelagic zone (Hoffmann et al. 2013; Schmiedeskamp et al. 2021), though CH<sub>4</sub> emissions might be highest in the center of small waterbodies due to ebullition (Matveev et al. 2016; Schmiedeskamp et al. 2021). In larger waterbodies, differences in gas transfer velocity across space might also lead to variability in dissolved CO<sub>2</sub> and CH<sub>4</sub> concentrations (Schilder et al. 2013). In small, shallow waterbodies, there is less space in which physical, chemical, and biotic drivers of CO<sub>2</sub> and CH<sub>4</sub> concentration can vary, and thus spatial variability of CO<sub>2</sub> and CH<sub>4</sub> in ponds and shallow lakes might be relatively less important than in larger, deeper waterbodies. However, if there is substantial spatial variability in CO<sub>2</sub> and CH<sub>4</sub> concentration in ponds and shallow lakes, sampling schemes that only measure from a single location in the waterbody are likely to misestimate concentration or emission.

Temporal variability in dissolved gas concentrations and diffusive fluxes in larger lentic systems exists across diel (Podgrajsek et al. 2014, 2015; Sieczko et al. 2020), weekly (Colas et al. 2020; Waldo et al. 2021), seasonal (Natchimuthu et al. 2017; Wiik et al. 2018; Paranaíba et al. 2021), and annual time frames (Finlay et al. 2019; Colas et al. 2020). Small lentic systems have similar temporal variability to larger lentic systems (Torgersen and Branco 2008; Huotari et al. 2009; Rudberg et al. 2021), but less is known about the controls of this variability as research focus has been on quantifying the magnitude, rather than the drivers, of diel and seasonal CO<sub>2</sub> and CH<sub>4</sub> dynamics (Wik et al. 2016; Natchimuthu et al. 2017; Waldo et al. 2021). We anticipate temporal variability in CO<sub>2</sub> and CH<sub>4</sub> concentrations in shallow lentic systems is likely to be higher than in larger and deeper systems due to more frequent and extreme changes in chemical (e.g., nutrient loading events) and physical factors (e.g., mixing events) that might be linked with CO<sub>2</sub> and CH<sub>4</sub> production and consumption.

Although previous efforts have quantified the importance of spatial and temporal variability in dissolved CO<sub>2</sub> or CH<sub>4</sub> concentration in one or just a few waterbodies, a broader analysis considering many waterbodies across a broad geographic range is needed to determine the prevalence of spatial and temporal variability across systems and to identify possible relationships with environmental variables that might be useful for predicting the most variable systems. In this study, we examined dissolved CO<sub>2</sub> and CH<sub>4</sub> concentrations in 30 shallow lentic waterbodies (i.e., ponds and shallow lakes) across temperate regions of Europe and North America. We aimed to:

(1) identify the main predictors of CO<sub>2</sub> and CH<sub>4</sub> concentrations for shallow waterbodies over a wide geographic scale; (2) quantify the spatial and temporal variability of dissolved CO<sub>2</sub> and CH<sub>4</sub> concentrations in these waterbodies, and determine how limited sampling in space and time might lead to misestimation of mean dissolved CO<sub>2</sub> and CH<sub>4</sub> concentrations; (3) identify waterbody characteristics that can be used to predict systems that are likely to have high spatial and temporal variability in CO<sub>2</sub> and CH<sub>4</sub> concentrations. This work is an important step in advancing our understanding of lentic CO<sub>2</sub> and CH<sub>4</sub> emissions, moving from identification of global emission patterns to reducing confidence intervals and uncertainty associated with these patterns (Downing 2009), helping to reduce uncertainty in global CH<sub>4</sub> emissions estimates.

## Methods

### Sampling locations and scheme

We sampled 30 ponds and shallow lakes in temperate areas of Europe and North America in summer 2018 and 2019 (June–September, though four sampling events took place in October; Fig. 1). Although there are various definitions of ponds and shallow lakes (Biggs et al. 2005), here we use the following: ponds have < 5 ha surface area and < 5 m maximum depth, while shallow lakes have > 5 ha surface area and < 5 m maximum depth (Scheffer 2004; Richardson et al. 2022). The waterbodies we sampled all had permanent hydroperiods and sediment bottoms. They were located in urban parks, residential areas, forests, and agricultural areas. Dissolved gas sampling at each site was conducted on three occasions (except Mud Pond, which was only sampled twice), spread across 61.7 d on average ( $\pm 25.6$  SD), ranging from 33 to 128 d between the first and last sampling date.

We measured waterbody surface area, perimeter, fetch, maximum depth, dissolved organic carbon (DOC), total phosphorus (P), dissolved P concentration, chlorophyll *a* (Chl *a*), conductivity, pH, Secchi depth, emergent plant cover, submerged plant cover, floating plant cover, and the presence or absence of fish. In some waterbodies, these environmental variables were measured once, while at other waterbodies we took the mean value from multiple sample dates. Chemical samples (i.e., DOC, total P, dissolved P, Chl *a*) were characterized using a variety of techniques, employing standard methods in the laboratory that collected the samples (Supporting Information Table S1). Not all variables were measured in all waterbodies or on all sampling occasions (Table 1; Supporting Information Table S1). However, for systems where chemical samples were collected on multiple occasions, within site variability was negligible compared to between site variability.

To test relationships among environmental variables, we used Pearson correlations (Supporting Information Table S2). Prior to estimating correlations and regressions, we checked whether the data distribution for each variable best fit a normal

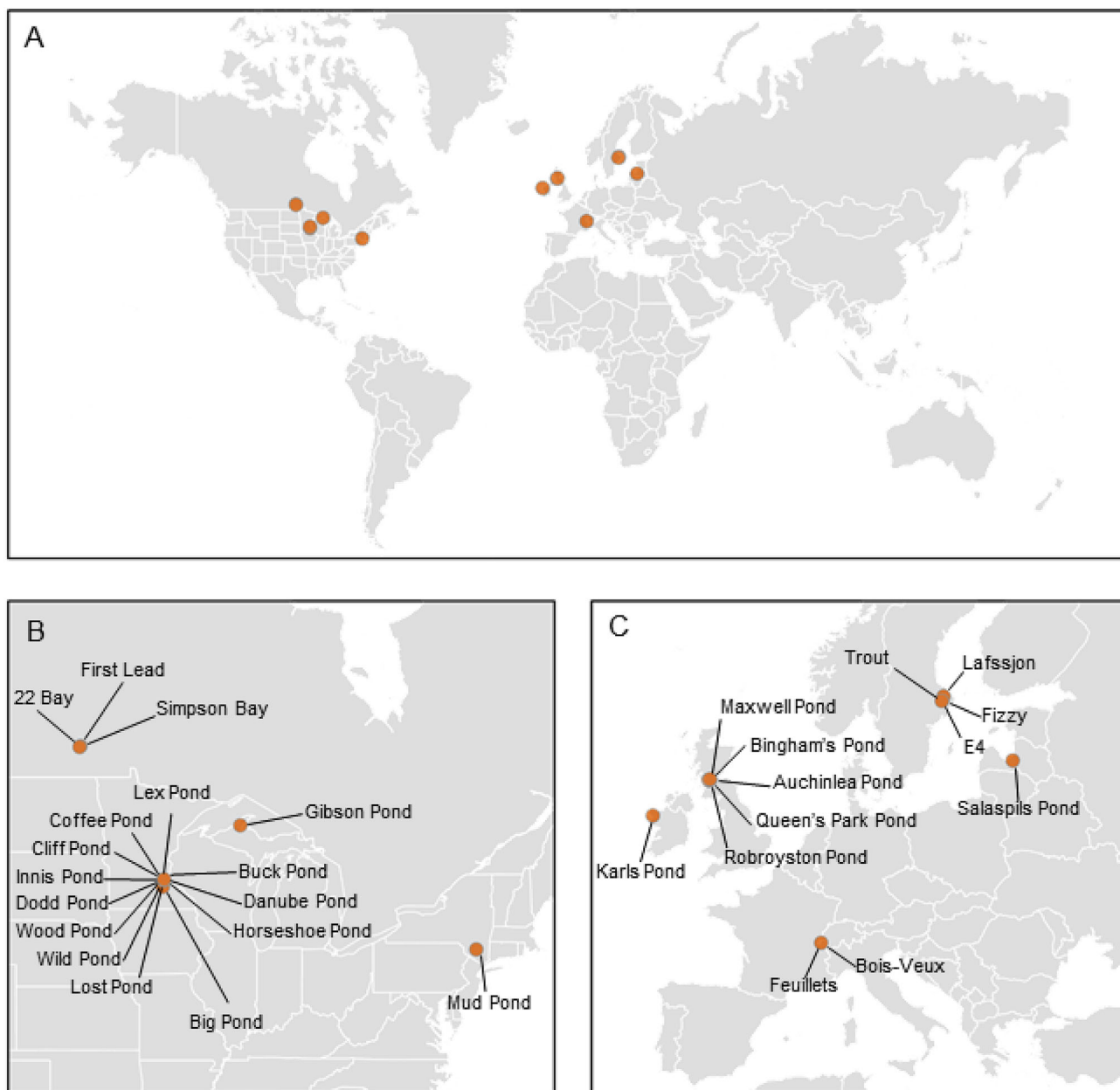
or lognormal distribution using the *fitdistrplus* package (Delignette-Muller et al. 2015), and made necessary transformations (Table 1). All statistical tests were conducted in R Statistical Software (R Core Team 2014) and we considered the results of statistical tests to be nominally significant (i.e., indicative of relationships that might be useful for explaining variation in the data) when  $p \leq 0.05$ .

### Dissolved gas sampling, analysis, and calculation of partial pressures

Gas sampling was conducted in the same way at all waterbodies, with samples collected from three locations in each waterbody on three occasions (in Gibson Pond and Mud Pond, samples were only collected from the waterbody center). On each sampling date, both air and dissolved gas samples were collected. Air samples ( $n = 2$ ) were collected from  $\sim 0.5$  m above the water surface in the center of the waterbody using syringes that were flushed with air three times prior to sample collection. Air samples were injected into pre-evacuated 12-mL glass exetainers (LabCo Limited). Dissolved gas concentrations were determined using a headspace equilibration technique (McAuliffe 1971; Holgerson 2015; Aho and Raymond 2019), and the headspace samples were stored in pre-evacuated glass exetainers. Two samples were collected from the waterbody center, and an additional sample was collected from each of two site margins (i.e., locations on opposite ends of the waterbody). Samples from the waterbody center were considered as technical replicates, and the average CO<sub>2</sub> and CH<sub>4</sub> concentration of these two samples was used in statistical analyses (we tested variability between the technical replicates as described later in the methods). All dissolved gas samples were collected from surface water by filling a syringe at < 15 cm depth. The temperature of both air and water was measured during sample collection. Atmospheric pressure was determined by the elevation of the waterbody above sea level.

Gas samples were analyzed at the Yale Analytical and Stable Isotope Center using a Shimadzu GC 2014 or at the University of Stirling using a Hewlett Packard GC 5890 Series II. Both instruments were equipped with a flame ionization detector for measuring CH<sub>4</sub>. Sample CO<sub>2</sub> and CH<sub>4</sub> concentrations were determined by comparing sample peak area against a standard curve of the peak areas of different concentrations of external standards. Dissolved CO<sub>2</sub> and CH<sub>4</sub> concentrations were then calculated for each sample following Henry's law and the ideal gas law using constants determined by Weiss (1974) and Wiesenburg and Guinasso (1979).

We converted dissolved gas concentrations to partial pressures ( $pX$ ;  $\mu\text{atm}$ ) using the following equation as presented by Aho and Raymond (2019) where  $[X]$  is the dissolved gas concentration ( $\mu\text{mol L}^{-1}$ ) and  $K_{H,X}$  is Henry's law solubility constant ( $\text{mol L}^{-1} \text{atm}^{-1}$ ) for CO<sub>2</sub> (Weiss 1974) or CH<sub>4</sub> (Wiesenburg and Guinasso 1979) given the temperature the water sample was collected:



**Fig. 1.** Locations of the 30 waterbodies sampled in this study (A), with panels showing location of waterbodies in North America (B) and Europe (C).

$$pX = \frac{[X]}{K_{h,x}} \quad (1)$$

We elected to present gas partial pressures to allow for simple prediction of whether a given location in a waterbody on a specific date is likely to be a source ( $pX >$  atmospheric  $[X]$ ), or sink ( $pX <$  atmospheric  $[X]$ ) of  $\text{CO}_2$  or  $\text{CH}_4$ .

#### 51 Environmental variables related to $p\text{CO}_2$ and $p\text{CH}_4$

52 We used both univariate and multivariate approaches to identify the best predictors and models of  $p\text{CO}_2$  and  $p\text{CH}_4$  as some environmental variables had low sample sizes (Table 1). We used

univariate linear regressions to identify the strength of the relationship between each chemical, physical, and biological variable measured and mean summer (all gas samples per waterbody)  $p\text{CO}_2$  and  $p\text{CH}_4$  for shallow lentic systems across a broad geographic range. Before calculating regressions, we checked distributions of  $p\text{CO}_2$  and  $p\text{CH}_4$ , again using the *fitdistrplus* package; the mean of all  $p\text{CO}_2$  values was normally distributed, while the mean of all  $p\text{CH}_4$  values was log normally distributed. We excluded Secchi depth from our analyses as it was strongly correlated with several other variables (maximum depth, DOC, total P, Chl *a*) and in several instances Secchi depth was unmeasurable as it was greater than waterbody maximum depth.



**Table 1.** Characteristics of 30 ponds and shallow lakes in temperate areas of Europe and North America in the summers of 2018 and 2019 sampled as part of this study. Waterbody-specific values can be accessed in the data file available online.

Characteristic	<i>n</i>	Data distribution	Mean	Median	Range
Latitude (°N)	30	Log-normal	49.95	46.64	41.69–60.02
Surface area (m <sup>2</sup> )	30	Log-normal	305,240	6227	180–8,230,000
Perimeter (m)	30	Log-normal	898	403	58–11,070
Fetch (m)	30	Log-normal	325.4	177.5	20.0–3190.0
Max depth (m)	30	Log-normal	1.6	1.3	0.6–4.8
Dissolved organic carbon (mg L <sup>-1</sup> )	28	Log-normal	10.8	7.8	4.8–32.5
Total phosphorus (μg L <sup>-1</sup> )	10	Log-normal	100.9	16.4	3.0–294.0
Dissolved phosphorus (μg L <sup>-1</sup> )	11	Log-normal	63.8	22.5	13.7–236.6
Chl <i>a</i> (μg L <sup>-1</sup> )	20	Log-normal	32.7	22.5	7.2–97.0
Conductivity (μs cm <sup>-1</sup> )	29	Log-normal	397.1	270.6	8.0–1722.0
pH	29	Normal	7.7	7.5	4.5–9.3
Secchi depth (m)	16	Normal	0.89	0.81	0.05–1.83
Emergent cover (% area)	24	Normal	12	10	0–40
Submerged cover (% area)	24	Normal	43	50	0–100
Floating cover (% area)	27	Normal	26	10	0–100
Fish (presence/absence)	30		15 present, 15 absent		

Next, we used multiple linear regression models, stepwise modeling, and an information theoretic model selection approach to determine the best-approximating model to describe mean  $p\text{CO}_2$  and  $p\text{CH}_4$ . The base model included variables measured in  $n \geq 28$  waterbodies. As several variables measured were strongly correlated with each other, we selected the variable with the largest sample size to include in the model, or if sample size was the same, we selected the variable that was significant in univariate regressions. Thus, the base model consisted of the following fixed effects: maximum depth, pH, DOC, fish presence, and one of surface area, perimeter, or fetch. Including DOC in all models slightly reduced our sample size as it was not measured in two waterbodies, but we elected to include it due to past evidence it is linked with aquatic  $\text{CO}_2$  and  $\text{CH}_4$  cycling (Deemer and Holgeron 2021; Peacock et al. 2021). For  $p\text{CO}_2$ , the base model included the following fixed effects: fetch, maximum depth, DOC, pH, and fish presence. We compared all combinations of fixed effects in this model by calculating Akaike information criterion scores corrected for small sample sizes (AICc) via the dredge function in the *MuMin* package (Barton 2020). We considered the best-approximating model to have the lowest AICc value, and considered models within 2  $\Delta\text{AICc}$  ( $\Delta\text{AICc}$  being the difference between the best-approximating and lower-ranked models) to be well supported (Burnham and Anderson 2002). We report models within 2  $\Delta\text{AICc}$  but do not interpret effects from those containing uninformative parameters (Arnold 2010). If the best-approximating model contained imprecisely estimated covariate effects (i.e., the ratio of the estimated effect to standard error was  $< \sim 2$ ), we only interpreted meaningful effects and advanced well estimated effects to subsequent modeling stages. To this model we then

iteratively added Chl *a*, % floating cover, and % emergent cover (at the cost of reduced df) to see if their inclusion would reduce AICc (recalculated for the inclusion of each new variable owing to changing sample sizes). We repeated this same process for  $p\text{CH}_4$ , replacing fetch with surface area, as surface area had a higher  $R^2$  than univariate models of perimeter or fetch. For all models, the fixed effects were scaled and fluxes log transformed in order for models to converge. Neither total P nor dissolved P were included in mixed effect model comparison due to their relatively small sample sizes.

### Spatial variability in $p\text{CO}_2$ and $p\text{CH}_4$

To determine the importance of spatial variability and sampling location within a waterbody, we considered the degree to which collecting samples from a single location in a waterbody might misestimate waterbody mean  $p\text{CO}_2$  or  $p\text{CH}_4$  from three sample locations using a bootstrap regression approach. We built the bootstrap model to randomly select a  $p\text{CO}_2$  or  $p\text{CH}_4$  value from a single sampling location in the waterbody on a given date as the response variable and the waterbody mean  $p\text{CO}_2$  and  $p\text{CH}_4$  on that date as the independent variable. We ran 1000 iterations of this model. We did not include waterbody as a random effect in our model despite repeated sampling as it prevented various iterations of the model from converging. Although exclusion of this random effect might be problematic when constructing a model with the goal of most accurately quantifying an  $R^2$  and  $p$ -value, our goal here was to quantify  $\beta$ , or the slope of the regression model. This  $\beta$  value is unlikely to be altered in such a magnitude to influence our interpretation of the model results regardless of the inclusion of the random effect.

1 We calculated a potential misestimate of waterbody  $p\text{CO}_2$   
 2 or  $p\text{CH}_4$  using the 95% confidence interval of slopes estimated  
 3 in the bootstrap regression (Eq. 2).

$$4 \text{ Potential \% Misestimate} = \frac{|2.5\% \text{ Quantile} - 1| + |97.5\% \text{ Quantile} - 1|}{2} \times 100$$

7 (2)

8 The calculated potential misestimate indicates by how much  
 9 the mean  $p\text{CO}_2$  or  $p\text{CH}_4$  of the water body might be misestimated  
 10 by sampling from a single location in the water body on a given  
 11 sampling event. It can be interpreted as the 95% likelihood of a single  
 12 sample location in the waterbody being within  $X\%$  of the mean  
 13 waterbody  $p\text{CO}_2$  or  $p\text{CH}_4$  on that sampling date.

14 Before testing for relationships between environmental variables  
 15 and spatial variability of  $p\text{CO}_2$  or  $p\text{CH}_4$ , we determined  
 16 whether variability (as standard deviation [SD]) among samples  
 17 collected over space was greater than variability of the center  
 18 technical replicates, in effect testing whether any spatial variability  
 19 we measured was greater than pure error. In over 85% of  
 20 the samples for both  $p\text{CO}_2$  and  $p\text{CH}_4$ , the variability in center  
 21 replicates was less than variability across the three sampling  
 22 locations in the waterbody (71 out of 84 for  $p\text{CO}_2$  and 74 out  
 23 of 84 for  $p\text{CH}_4$ ; Supporting Information Fig. S2) when samples  
 24 were collected at multiple locations. When technical variability  
 25 was higher than variability across sampling locations within a  
 26 waterbody, it was typically when mean  $p\text{CO}_2$  or  $p\text{CH}_4$  was low  
 27 (and thus any variability among technical replicates would  
 28 appear greater) or the spatial variability was low relative to the  
 29 mean. As such, our sampling approach accurately reflects spatial  
 30 variability and is not instead driven by pure error.

31 To estimate the relative spatial variability of  $p\text{CO}_2$  or  $p\text{CH}_4$   
 32 in waterbodies, we used residuals of the linear relationship  
 33  $\log(\text{SD}_{p\text{CO}_2 \text{ or } p\text{CH}_4}) \sim \log(\text{mean}_{p\text{CO}_2 \text{ or } p\text{CH}_4})$  for each waterbody  
 34 on each sampling day (Supporting Information Fig. S2). We  
 35 then used univariate linear mixed effects models to test the  
 36 relationship between each waterbody characteristic and the  
 37  $p\text{CO}_2$  or  $p\text{CH}_4$  residual, with waterbody as a random effect.  
 38 Models were constructed using the *lme4* and *lmerTest* packages  
 39 (Bates et al. 2015; Kuznetsova et al. 2017). Conditional and  
 40 marginal  $R^2$  values for each model were calculated using the  
 41 *sjstats* package (Lüdtke 2021). Two ponds (Gibson Pond and  
 42 Mud Pond) were excluded from the spatial variability analysis  
 43 as sampling was only conducted in the waterbody center. We  
 44 also conducted a multivariate analysis to identify the best combination  
 45 of variables to use to identify systems that might be  
 46 more or less variable following the same approach described  
 47 previously for mean  $p\text{CO}_2$  or  $p\text{CH}_4$ , but instead using linear  
 48 mixed effects models with the addition of waterbody as a random  
 49 effect to account for repeated measures.

### 52 Temporal variability in $p\text{CO}_2$ and $p\text{CH}_4$

53 We used a similar bootstrap approach as described for spatial  
 54 variability to quantify the uncertainty in mean  $p\text{CO}_2$  and

$p\text{CH}_4$  associated with sampling each waterbody only once. In  
 the bootstrap regression we used the waterbody mean  $p\text{CO}_2$   
 or  $p\text{CH}_4$  on a randomly selected date as the response variable  
 and mean of all  $p\text{CO}_2$  or  $p\text{CH}_4$  values from three sampling  
 dates in that waterbody as the independent variable. Potential  
 misestimate of  $p\text{CO}_2$  or  $p\text{CH}_4$  is determined using Eq. 2.  
 It can be interpreted as the 95% likelihood of a single  $p\text{CO}_2$   
 or  $p\text{CH}_4$  sampling event being within  $X\%$  of the mean  $p\text{CO}_2$   
 or  $p\text{CH}_4$  of three summer sampling events. We repeated the  
 bootstrap approach a third time, using a random, single sample  
 from each waterbody compared against the mean of all  
 samples collected in space and time to calculate the potential  
 misestimate of mean summer  $p\text{CO}_2$  or  $p\text{CH}_4$  from a single  
 grab sample.

We used similar univariate and multivariate approaches  
 to identify predictors of temporal variability as described  
 previously for mean  $p\text{CO}_2$  or  $p\text{CH}_4$  and spatial variability in  
 $p\text{CO}_2$  or  $p\text{CH}_4$ , but here we calculated residuals for  $p\text{CO}_2$   
 and  $p\text{CH}_4$  for each waterbody using the mean  $p\text{CO}_2$  and  
 $p\text{CH}_4$  from each of the samples collected per waterbody on  
 each sampling date (Supporting Information Fig. S3) and  
 again used multiple linear regression. One pond (Mud Pond)  
 was only sampled twice and was therefore excluded from  
 these calculations.

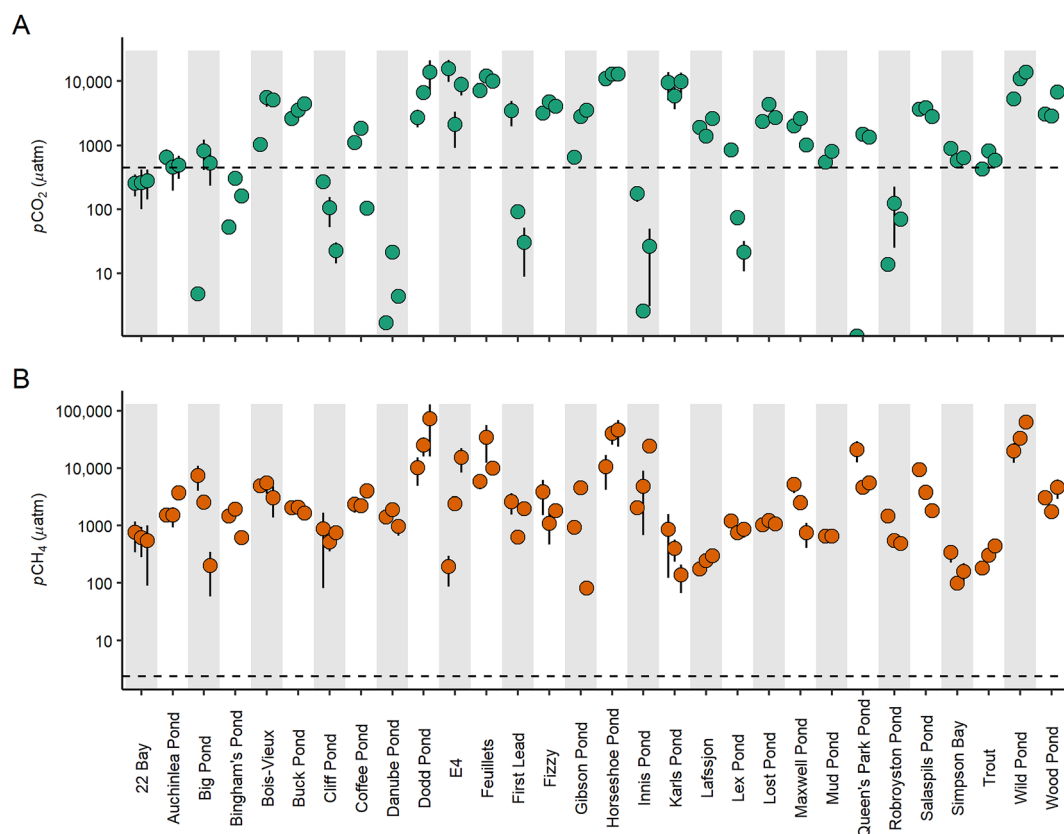
## Results

### Waterbody characteristics

The sampled ponds and shallow lakes had a large range of  
 physical, chemical, and biological characteristics (Table 1).  
 There were several significant correlations between these  
 characteristics (Supporting Information Table S2), including  
 strong positive correlations between perimeter, fetch, and  
 surface area ( $r \geq 0.90$ ,  $p < 0.01$ ). Notably, surface area and  
 maximum depth were not correlated ( $r = -0.07$ ;  $p = 0.71$ ;  
 $df = 28$ ). The two largest waterbodies (22 Bay and Simpson  
 Bay;  $> 100,000 \text{ m}^2$ ) and three smallest (E4, Fizzy, Karls Pond;  
 $< 1000 \text{ m}^2$ ) all have a similar maximum depth (0.64–1.25 m).  
 The system with the greatest maximum depth was Lost Pond  
 (4.8 m) which has a surface area ( $6354 \text{ m}^2$ ) similar to the  
 dataset median ( $6227 \text{ m}^2$ ).

### Environmental variables related to $p\text{CO}_2$ and $p\text{CH}_4$

On average, waterbodies had mean  $p\text{CO}_2$  ( $3094 \pm 3576$   
 $\mu\text{atm}$ ; mean  $\pm$  SD; Fig. 2a) nearly  $7\times$  higher than the mean  
 $p\text{CO}_2$  of air samples ( $446.0 \pm 40.0 \mu\text{atm}$ ) indicative of super-  
 saturation and net release of  $\text{CO}_2$  to the atmosphere. Six  
 waterbodies had mean  $p\text{CO}_2$  below atmospheric concentra-  
 tion on all three sampling events indicating they were  
 net  $\text{CO}_2$  sinks. Eight waterbodies had variable source-sink  
 behavior across sampling dates, and several had variable  
 source-sink behavior at different locations in the system on  
 individual sampling dates.  $p\text{CH}_4$  ranged across several orders  
 of magnitude from a low of  $199.5 \mu\text{atm}$  in Simpson Bay to a



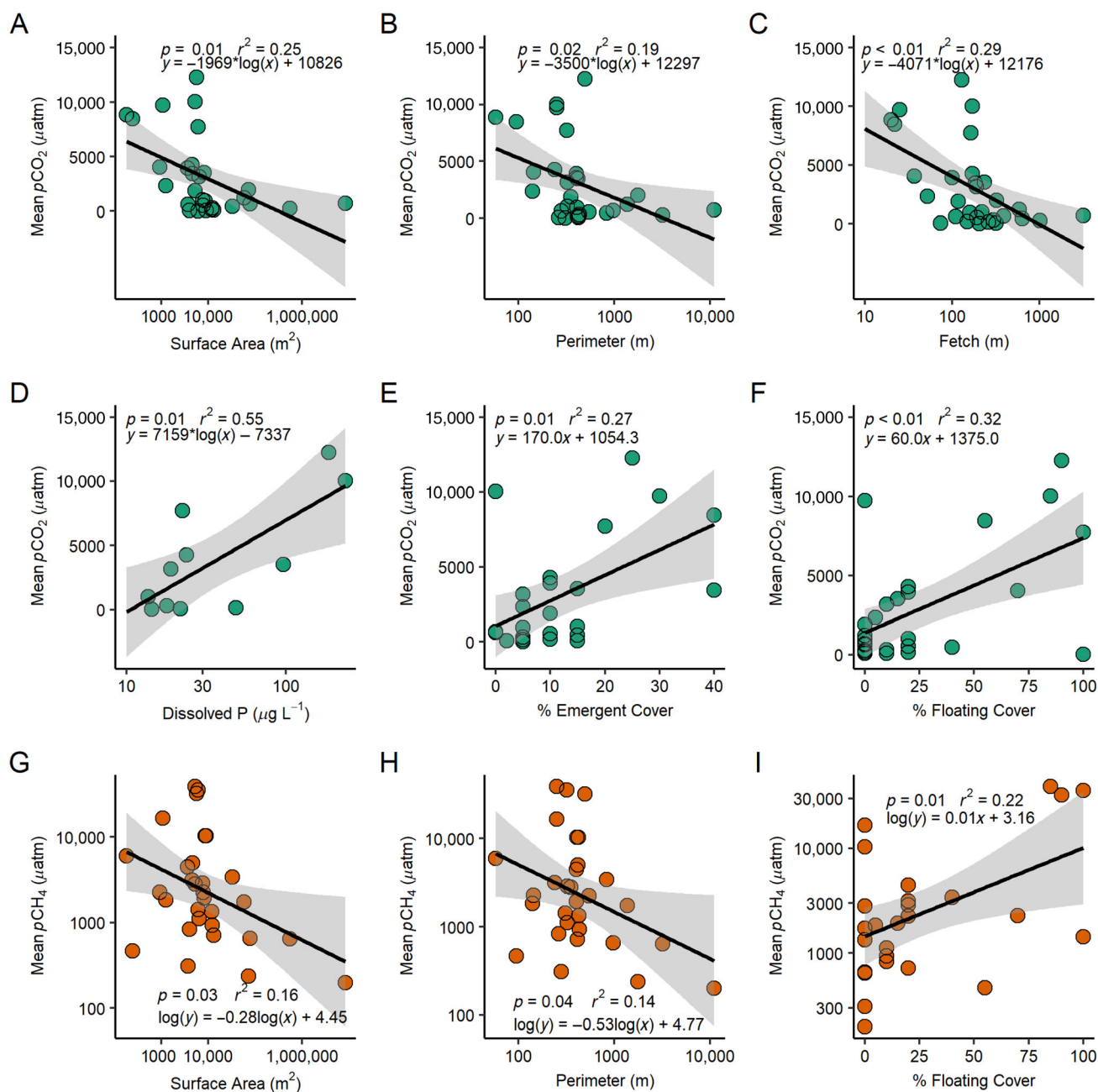
**Fig. 2.** Partial pressures of carbon dioxide (**A**;  $p\text{CO}_2$ ) and methane (**B**;  $p\text{CH}_4$ ) in surface water of Measurements were made in 30 ponds and shallow lakes in temperate areas of Europe and North America in the summers of 2018 and 2019. Each point indicates mean gas partial pressure in a single waterbody on a single sampling date. The error bars represent the SD in  $p\text{CO}_2$  or  $p\text{CH}_4$  in space on that sampling date. Dashed lines indicate mean atmospheric gas concentration across all sampling events with points above the line indicative of gas release to the atmosphere and points below indicative of uptake by the waterbody. In cases where error bars are hidden, the SD is very small (there is no SD for Gibson Pond or Mud Pond as samples were collected from a single location in these waterbodies).

high of  $38,803 \mu\text{atm}$  in Wild Pond, though all systems had partial pressures of  $\text{CH}_4$  (mean  $p\text{CH}_4 = 6350 \pm 10,578 \mu\text{atm}$ ; Fig. 2B) higher than the atmosphere ( $2.43 \pm 0.66 \mu\text{atm}$ ) across all sampling dates and sampling locations. Generally, waterbodies with high mean  $p\text{CO}_2$  had high mean  $p\text{CH}_4$  (Supporting Information Fig. S1).

We identified several physical and biological variables that related to  $p\text{CO}_2$  and  $p\text{CH}_4$  (Supporting Information Table S9; Fig. 3). Waterbodies with smaller surface areas had higher  $p\text{CO}_2$  ( $r^2 = 0.25$ ;  $p = 0.01$ ;  $\text{df} = 28$ ) and  $p\text{CH}_4$  ( $r^2 = 0.16$ ;  $p = 0.03$ ;  $\text{df} = 28$ ) than those with larger surface areas. The percent area of the waterbody covered in floating vegetation positively related to both  $p\text{CO}_2$  ( $r^2 = 0.32$ ;  $p < 0.01$ ;  $\text{df} = 25$ ) and  $p\text{CH}_4$  ( $r^2 = 0.22$ ;  $p = 0.01$ ;  $\text{df} = 25$ ). Emergent vegetation cover was positively related to  $p\text{CO}_2$  ( $r^2 = 0.27$ ;  $p = 0.01$ ;  $\text{df} = 22$ ) but not  $p\text{CH}_4$ . The variable that mostly strongly predicted  $p\text{CO}_2$  was dissolved P concentration ( $r^2 = 0.55$ ;  $p = 0.01$ ;  $\text{df} = 9$ ), which had a positive relationship, though the sample size was relatively low ( $n = 11$ ) compared to most other measures. Fish presence related to both  $\text{CO}_2$  and  $\text{CH}_4$  concentrations:  $p\text{CO}_2$  was

almost four times higher in fishless systems ( $4765 \pm 4272 \mu\text{atm CO}_2$ ) than in those with fish ( $1423 \pm 1502 \mu\text{atm CO}_2$ ;  $p < 0.01$ ;  $\text{df} = 28$ ), and  $p\text{CH}_4$  was nearly five times greater in fishless systems ( $10,545 \pm 13,688 \mu\text{atm CH}_4$ ) relative to those with fish ( $2156 \pm 2597 \mu\text{atm CH}_4$ ;  $p = 0.03$ ;  $\text{df} = 28$ ).

The best-approximating multivariate model to describe waterbody mean  $p\text{CO}_2$  included DOC ( $\beta = 0.28$ ;  $\text{SE} = 0.02$ ), fish presence ( $\beta = -0.72$ ;  $\text{SE} = 0.29$ ), and pH, but the pH effect was not well estimated ( $\beta = -0.27$ ;  $\text{SE} = 0.14$ ;  $R^2 = 0.30$ ;  $p < 0.01$ ;  $\text{df} = 25$ ; Supporting Information Table S3). The addition of various primary producers did not improve the model's ability to predict mean  $p\text{CO}_2$  (Supporting Information Table S3). The best-approximating multivariate model to describe waterbody mean  $p\text{CH}_4$  was fish presence alone ( $\beta = -0.57$ ;  $\text{SE} = 0.22$ ; Supporting Information Table S4). The addition of Chl *a* did not improve the model, but the addition of floating and submerged plant cover did (Supporting Information Table S4). Floating plant cover was positively associated with  $p\text{CH}_4$  ( $\beta = 0.25$ ;  $\text{SE} = 0.11$ ) as was submerged plant cover ( $\beta = 0.27$ ;  $\text{SE} = 0.12$ ).



**Fig. 3.** Relationships between mean partial pressures of carbon dioxide ( $p\text{CO}_2$ ) and waterbody (A) surface area, (B) perimeter, (C) fetch, (D) dissolved phosphorus concentration, (E) emergent cover, and (F) floating cover, and between partial pressures of methane ( $p\text{CH}_4$ ) and waterbody (G) surface area, (H) perimeter, and (I) floating cover. Only relationships with  $p \leq 0.05$  shown, other relationships with  $p > 0.05$  in Supporting Information Table S9. Measurements were made in 30 ponds and shallow lakes in temperate areas of Europe and North America in the summers of 2018 and 2019.

### Spatial variability in $p\text{CO}_2$ and $p\text{CH}_4$

Bootstrap regressions indicated that randomly sampling from a single location in small waterbodies results in low (13%) misestimates in  $p\text{CO}_2$  (Table 2). This relatively low spatial variability in  $p\text{CO}_2$  was further evidenced by the lack of any environmental variables that were significantly correlated with  $p\text{CO}_2$  residuals (Supporting Information Table S10). There was slightly more spatial variability in  $p\text{CH}_4$  (35%

potential misestimate in space; Table 2), and we found that spatial variability was negatively correlated with water depth ( $R^2 = 0.13$ ,  $p < 0.01$ ,  $n = 84$ ; Fig. 4A), and positively correlated with Chl *a* concentration ( $R^2 = 0.08$ ,  $p = 0.05$ ,  $n = 57$ ; Fig. 4B) and conductivity ( $R^2 = 0.08$ ,  $p = 0.02$ ,  $n = 81$ ; Fig. 4C). Using a multivariate approach, the best model to approximate  $p\text{CO}_2$  variability in space was the null model ( $R^2 = 0.00$ ,  $p = 0.17$ ,  $n = 81$ ; Supporting Information Table S5), while the best



**Table 2.** Results of bootstrap regressions ( $n = 1000$  iterations) of randomly sampled partial pressures of carbon dioxide ( $p\text{CO}_2$ ) or methane ( $p\text{CH}_4$ ) in space and time relative to mean  $p\text{CO}_2$  or  $p\text{CH}_4$ . “Space” refers to selecting a  $p\text{CO}_2$  or  $p\text{CH}_4$  value from a single location in the waterbody relative to the waterbody mean  $p\text{CO}_2$  or  $p\text{CH}_4$  on a given date, “Time” refers to randomly selecting waterbody mean  $p\text{CO}_2$  or  $p\text{CH}_4$  on a single date relative to the seasonal mean  $p\text{CO}_2$  or  $p\text{CH}_4$ , and “Time and Space” refers to selecting a single  $p\text{CO}_2$  or  $p\text{CH}_4$  sample as representative of the seasonal mean. Potential misestimate is calculated as described in Eq. 3. Measurements were made in 30 ponds and shallow lakes in temperate areas of Europe and North America in the summers of 2018 and 2019.

	$R^2$	$p$ -value	Mean intercept	Intercept 95% CI	Mean slope	Slope 95% CI	Min. slope	Max. slope	Potential % misestimate
Space $p\text{CO}_2$	0.92	< 0.01	-2.17	-217.2 to 245.8	1.00	0.87-1.13	0.83	1.16	13
Space $p\text{CH}_4$	0.90	< 0.01	-37.37	-1302 to 1370	1.00	0.63-1.33	0.56	1.44	35
Time $p\text{CO}_2$	0.90	< 0.01	7.75	-396.2 to 410.3	0.99	0.73-1.25	0.61	1.33	26
Time $p\text{CH}_4$	0.93	< 0.01	3.91	-2123 to 1829	0.99	0.39-1.68	0.35	1.76	64.5
Time and Space $p\text{CO}_2$	0.92	< 0.01	-0.19	-403.0 to 423.3	1.00	0.72-1.31	0.62	1.50	44
Time and Space $p\text{CH}_4$	0.87	< 0.01	-17.96	-3015 to 2236	0.99	0.36-2.03	0.21	2.42	83.5

model to describe variability of  $p\text{CH}_4$  in space was maximum depth alone ( $\beta = -0.13$ ;  $\text{SE} = 0.05$ ; Supporting Information Table S6). Primary producers did not improve either model (Supporting Information Tables S5, S6).

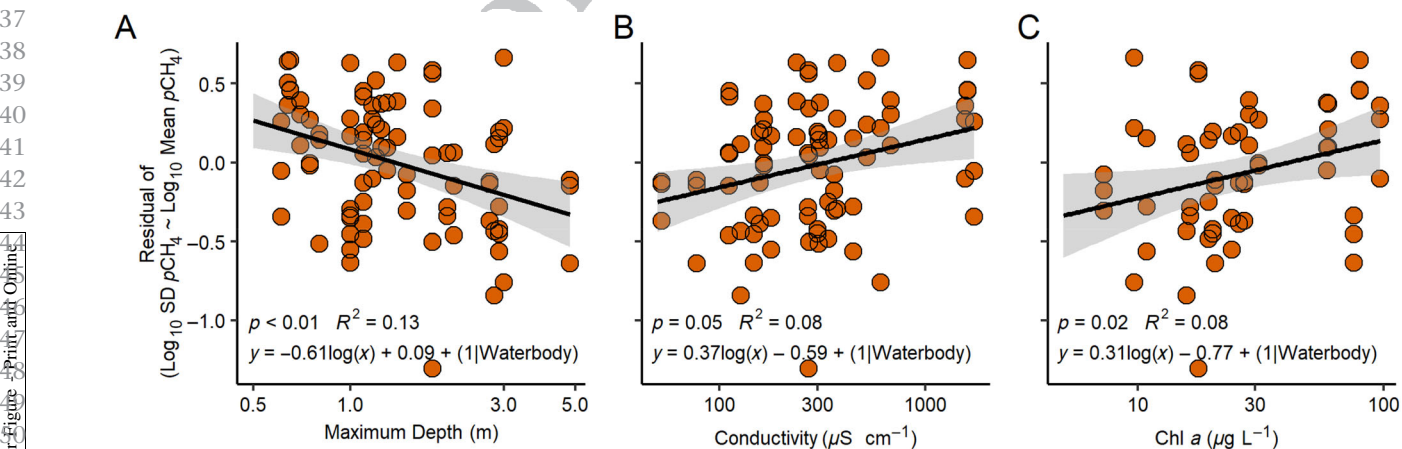
#### Temporal variability in $p\text{CO}_2$ and $p\text{CH}_4$

The variability of dissolved gas concentrations was greater in time than space. Bootstrap regressions reveal potential misestimation of summer mean  $p\text{CO}_2$  by up to 26% and  $p\text{CH}_4$  by up to 64.5% if sampling is only conducted on a single date (Table 2). Taken a step further, the potential misestimate increases to 44% for  $p\text{CO}_2$  and 83.5% for  $p\text{CH}_4$  (Table 2) if only a single sample from a random location in the waterbody on a single sampling event (the combined effects of spatial and temporal variability) is used to estimate mean summer  $p\text{CO}_2$  or  $p\text{CH}_4$ .

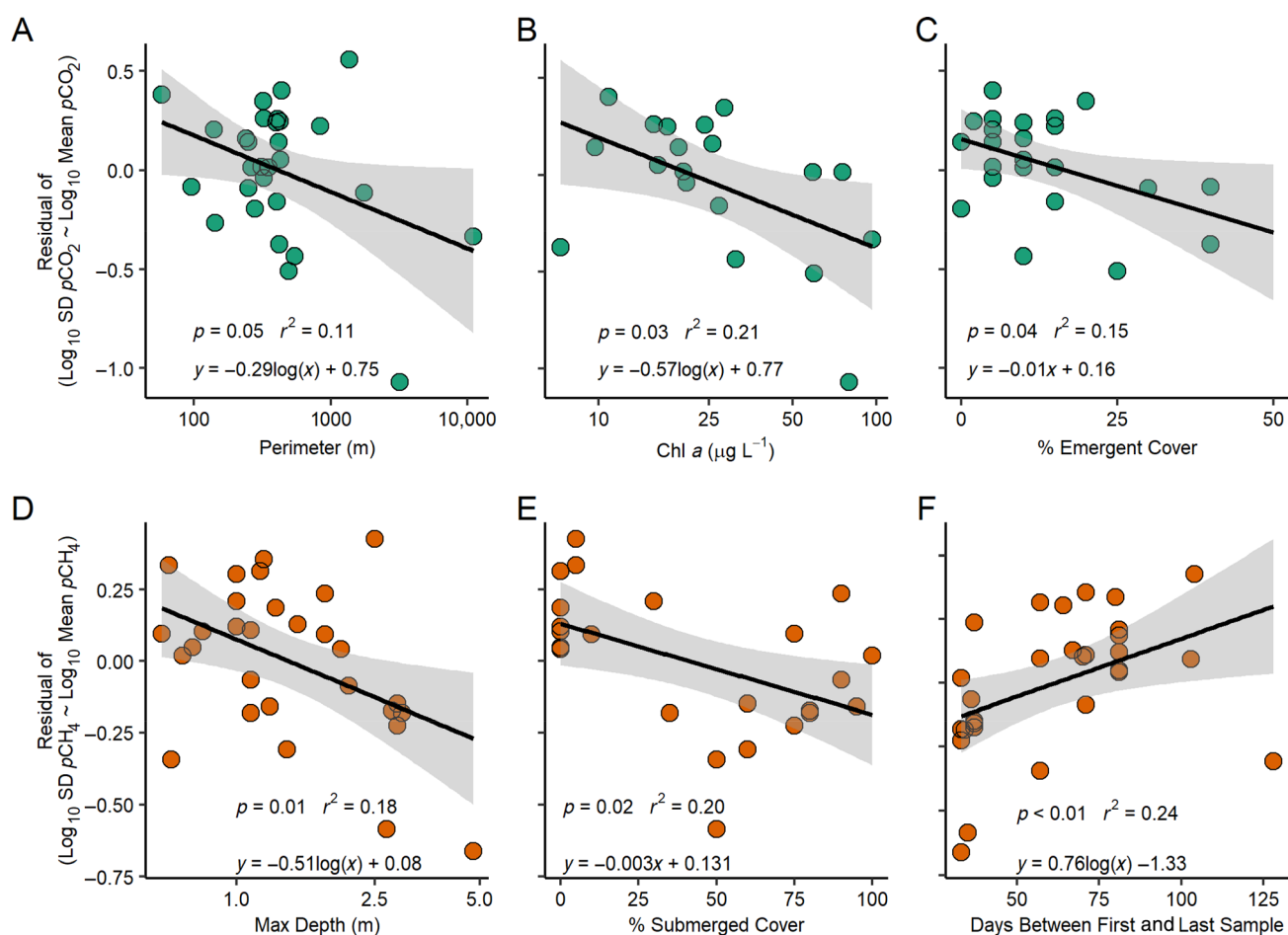
Temporal variability in  $p\text{CO}_2$  was negatively correlated with waterbody perimeter ( $r^2 = 0.11$ ,  $p = 0.05$ ;  $\text{df} = 27$ ; Fig. 5A),

Chl  $a$  concentration ( $r^2 = 0.21$ ,  $p = 0.03$ ;  $\text{df} = 17$ ; Fig. 5B), and percent emergent cover ( $r^2 = 0.15$ ,  $p = 0.04$ ;  $\text{df} = 21$ ; Fig. 5C). There was no relationship between temporal variability in  $p\text{CO}_2$  and the length of time between the first and last sampling event. Temporal variability of  $p\text{CH}_4$  decreased as the waterbody maximum depth ( $r^2 = 0.18$ ,  $p = 0.01$ ;  $\text{df} = 27$ ; Fig. 5D) and percent submerged cover increased ( $r^2 = 0.20$ ,  $p = 0.02$ ;  $\text{df} = 21$ ; Fig. 5E), and was positively correlated with sampling time frame ( $r^2 = 0.24$ ,  $p < 0.01$ ;  $\text{df} = 27$ ; Fig. 5F).

The multivariate model with the lowest AICc score for describing variability of  $p\text{CO}_2$  over time was perimeter alone, but the effect of perimeter was not well estimated ( $\beta = -0.11$ ;  $\text{SE} = 0.07$ ; Table S7). The addition of Chl  $a$  improved the null model, and Chl  $a$  was negatively correlated with variability of  $p\text{CO}_2$  over time ( $\beta = -0.21$ ;  $\text{SE} = 0.07$ ). Similarly, the addition of emergent cover improved the model and emergent cover was negatively associated with variability of  $p\text{CO}_2$  over time ( $\beta = -0.11$ ;  $\text{SE} = 0.05$ ). Inclusion of floating and submerged



**Fig. 4.** Relationships between spatial variability of the partial pressure of methane ( $p\text{CH}_4$ ) and maximum depth (A), conductivity (B), and Chl  $a$  (C). Only relationships with  $p \leq 0.05$  shown, other relationships with  $p > 0.05$  in Supporting Information Table S10. The (1|Waterbody) indicates inclusion of waterbody as a random effect in the model.  $R^2$  values shown are the marginal  $R^2$  of the model. Measurements were made in 30 ponds and shallow lakes in temperate areas of Europe and North America in the summers of 2018 and 2019.



**Fig. 5.** Relationships between temporal variability of the partial pressure of carbon dioxide ( $p\text{CO}_2$ ) and waterbody perimeter (A), Chl  $a$  (B), and % emergent cover (C) and between temporal variability of the partial pressure of methane ( $p\text{CH}_4$ ) and maximum waterbody depth (D), percent submerged cover (E), and the number of days between the first and last sample collected (F). Only relationships with  $p \leq 0.05$  shown, other relationships with  $p > 0.05$  in Supporting Information Table S11. Measurements were made in 30 ponds and shallow lakes in temperate areas of Europe and North America in the summers of 2018 and 2019.

plant cover did not improve the model (Supporting Information Table S7). For temporal variability of  $p\text{CH}_4$  the best-approximating model was maximum depth alone ( $\beta = -0.16$ ; SE = 0.04) and primary producers did not improve the model (Supporting Information Table S8).

## Discussion

Identifying drivers of  $\text{CO}_2$  and  $\text{CH}_4$  concentrations in small and shallow waterbodies is critical for accurate inclusion of these systems in global  $\text{CO}_2$  and  $\text{CH}_4$  budgets. Determining how these systems vary in space and time will guide targeted sampling and further reduce error in our global estimates, and thus improve accuracy in scaling. Here, we found a mix of source/sink behavior for  $p\text{CO}_2$  across waterbodies, sampling dates, and locations within the waterbody, whereas all waterbodies were supersaturated in  $\text{CH}_4$ . Both  $p\text{CO}_2$  and  $p\text{CH}_4$  spanned 4 orders of magnitude across the 30 waterbodies

representing a broad geographic range. We took advantage of this variability to identify relationships between physical, chemical, and biological parameters and  $\text{CO}_2$  and  $\text{CH}_4$  concentration and variability, providing important insight into which systems may be the most variable.

### Environmental variables related to $p\text{CO}_2$ and $p\text{CH}_4$

Despite our focus on shallow and relatively small systems, we still observed inverse relationships between waterbody size (i.e., surface area, fetch, perimeter) and  $p\text{CO}_2$  and  $p\text{CH}_4$  similar to relationships observed across a wider range of waterbody sizes (Holgerson and Raymond 2016; Deemer and Holgerson 2021). In smaller lentic systems it can be unclear whether the negative relationships between size and  $\text{CO}_2$  or  $\text{CH}_4$  concentrations are driven by physical processes or by chemical/biological drivers of  $\text{CO}_2$  or  $\text{CH}_4$  concentration that can co-vary with size. In our dataset there were correlations between waterbody size (i.e., surface area, perimeter, fetch)

1 and proxies of nutrient and organic matter loading (i.e., DOC,  
2 total P, dissolved P concentration), but only dissolved P  
3 predicted  $p\text{CO}_2$  and none of the chemical variables measured  
4 in this study predicted  $p\text{CH}_4$ . Together, this indicates that phys-  
5 ical and biological factors may have a greater effect than chemi-  
6 ical factors (or bulk chemical pools) on  $\text{CH}_4$  concentrations in  
7 small freshwater systems. This conclusion is supported by evi-  
8 dence from boreal lakes  $< 0.07 \text{ km}^2$  in Finland, where water col-  
9 umn stability and turbulent mixing in smaller systems were  
10 more important than total organic carbon (TOC) loading from  
11 the surrounding landscape in predicting  $\text{CH}_4$  (Kankaala  
12 et al. 2013), despite co-variance between lake size and TOC.

13 Dissolved P concentration had the strongest relationship  
14 with  $p\text{CO}_2$ , with highest  $p\text{CO}_2$  when dissolved P concentra-  
15 tion was highest (although dissolved P concentration was only  
16 measured in 11 waterbodies). Multivariate analysis included  
17 DOC and fish presence in the best-approximating model (dis-  
18 solved P was not included in multivariate analysis due to small  
19 sample size). We are unable to determine the underlying  
20 mechanisms behind these relationships but can suggest two  
21 non-mutually exclusive hypotheses. First, systems with high  
22 organic matter loading (whether from internal or external  
23 sources) are likely to have high rates of sediment respiration  
24 and release of  $\text{CO}_2$ , DOC, and dissolved P to the water col-  
25 umn. Second, groundwater and runoff derived dissolved P,  
26 DOC, and  $\text{CO}_2$  loaded to small lentic systems could be concu-  
27 rrent (Marcé et al. 2015; Peacock et al. 2019). Jensen et al.  
28 (2022) report a positive relationship between DOC and dis-  
29 solved  $\text{CO}_2$  concentration and a negative relationship between  
30  $\delta^{18}\text{O}$  (indicative of groundwater influence) and dissolved  $\text{CO}_2$   
31 concentration in small agricultural reservoirs, indicating the  
32 importance of runoff and groundwater in DOC loading and  
33  $\text{CO}_2$  production.

34 Vegetation can also regulate  $\text{CO}_2$  and  $\text{CH}_4$  in aquatic sys-  
35 tems (Bodmer et al. 2021; Bastviken et al. 2023). We found  
36 that the percent of the waterbody covered with floating vege-  
37 tation related positively to both  $p\text{CO}_2$  and  $p\text{CH}_4$  and the per-  
38 cent of the waterbody area covered with emergent vegetation  
39 was positively related to  $p\text{CO}_2$ . Emergent cover had a strong  
40 negative correlation with surface area, but the addition of  
41 emergent cover to the best-approximating model—which did  
42 not include waterbody area—improved the model, hinting  
43 that vegetation may be more important than surface area in  
44 regulating  $p\text{CO}_2$  in small waterbodies. On the other hand, the  
45 areal coverage of floating vegetation was not correlated with  
46 any other environmental variables measured (Supporting  
47 Information Table S2), indicating a clear effect where floating  
48 vegetation increased both  $p\text{CO}_2$  and  $p\text{CH}_4$ . Floating plants can  
49 reduce gas exchange between the water column and the atmo-  
50 sphere, preventing diffusion of  $\text{O}_2$  into the water column and  
51 allowing for a buildup of  $\text{CO}_2$  and  $\text{CH}_4$  (Goodwin et al. 2008;  
52 Kosten et al. 2016; Rabaey and Cotner 2022). Alternatively,  
53 floating plants can reduce  $\text{CH}_4$  concentrations in surface water  
54 via oxygen loss through their roots and by providing surface

area for methanotrophic bacteria. The balance of reduced dif- 55  
56 fusion due to physical obstruction with enhanced oxidation  
57 via root transfer ultimately dictates how floating vegetation  
58 will alter surface water  $\text{CH}_4$  concentrations. In this study, both  
59  $p\text{CO}_2$  and  $p\text{CH}_4$  increased as floating plant cover increased  
60 suggesting reduced gas transfer drove this pattern.

61 Fish presence was an important indicator of average  
62 waterbody  $p\text{CO}_2$  and  $p\text{CH}_4$ , which approximately four and  
63 five times higher, respectively, in fishless waterbodies com-  
64 pared to those with fish. Although there is evidence that fish  
65 can alter aquatic  $\text{CO}_2$  and  $\text{CH}_4$  cycling (Schindler et al. 1997;  
66 Atwood et al. 2013; Devlin et al. 2015) it is also possible that  
67 fish presence may simply correlate with other factors that reg-  
68 ulate  $p\text{CO}_2$  and  $p\text{CH}_4$  (e.g., anoxia-driven winter fish kills). We  
69 can conclude that fish presence is a useful variable to measure  
70 for predicting  $p\text{CO}_2$  and  $p\text{CH}_4$  in ponds and shallow lakes and  
71 more work to quantify how fish alter  $p\text{CO}_2$  and  $p\text{CH}_4$  is  
72 needed.  
73

#### 74 Spatial variability in $p\text{CO}_2$ and $p\text{CH}_4$

75 Results of this study support past evidence that spatial vari-  
76 ability in lentic systems  $< 10 \text{ km}^2$  may be important for accu-  
77 rate quantification of  $p\text{CH}_4$  (Wik et al. 2016; Natchimuthu  
78 et al. 2017), with the possibility of misestimating waterbody  
79 mean  $p\text{CH}_4$  by up to 35% if only one location in the  
80 waterbody is sampled. Spatial variability in  $p\text{CO}_2$  appears less  
81 important in these small waterbodies and accurate estimates  
82 of waterbody  $p\text{CO}_2$  can likely be made from a single location.

83 Only three variables related to the spatial variability of  $p\text{CH}_4$ ,  
84 and model selection indicates spatial variability of  $p\text{CH}_4$  is best  
85 described by the maximum depth of the system, with less vari-  
86 ability in deeper waterbodies. We expected the opposite: that  
87 deeper systems would be more spatially variable as littoral zones  
88 may have greater  $\text{CH}_4$  concentrations than deeper waters  
89 (Hofmann 2013; Schmiedeskamp et al. 2021). We can test  
90 whether basin shape is related to spatial variability in  $p\text{CH}_4$  using  
91 the ratio of surface area to maximum depth. Doing so, we found  
92 no relationship between  $p\text{CH}_4$  variability and this ratio (marginal  
93  $R^2 < 0.01$ ). An alternative explanation for the observed negative  
94 relationship between depth and spatial variability considers strat-  
95 ification dynamics, which can be associated with maximum  
96 depth (Holgerson et al. 2022). Deeper systems with stronger  
97 stratification may become anoxic in bottom waters, favoring  
98  $\text{CH}_4$  production, but potentially trapping this  $\text{CH}_4$  beneath the  
99 thermocline, with little exchange of  $\text{CH}_4$  with surface waters; in  
100 contrast, shallow waters may have both more horizontal and  
101 vertical mixing that could create more spatial heterogeneity in  
102  $\text{CH}_4$  concentration. Disruption of stratification is an important  
103 driver of spatial variability in  $\text{CH}_4$  concentrations in larger sys-  
104 tems (Paranaíba et al. 2018, 2021), and may be similarly impor-  
105 tant in small lentic systems.

106 Spatial variability in  $p\text{CH}_4$  increased with Chl *a* concentration  
107 and conductivity in univariate regressions, but in multivariate  
108 analysis, neither was included in the best model. Chl *a* may

1 indicate increased organic matter loading in some areas (with  
2 subsequent spatial variability in CH<sub>4</sub> production) or if produc-  
3 tion is synchronous throughout the waterbody, areas with  
4 anoxic conditions may favor slightly higher CH<sub>4</sub> production,  
5 again leading to relatively higher spatial variability than systems  
6 with lower planktonic primary production.

### 8 Temporal variability in pCO<sub>2</sub> and pCH<sub>4</sub>

9 The potential for misestimating pCO<sub>2</sub> or pCH<sub>4</sub> over time  
10 was nearly twice the potential of misestimating pCO<sub>2</sub> or pCH<sub>4</sub>  
11 in space and was more important for pCH<sub>4</sub> (64.5% potential  
12 misestimate) than for pCO<sub>2</sub> (26% potential misestimate). Con-  
13 sidering only univariate regressions, both pCO<sub>2</sub> and pCH<sub>4</sub>  
14 were associated with waterbody physical and biological char-  
15 acteristics, with greater variability in relatively smaller systems  
16 with less primary producer biomass. Temporal variability of  
17 pCO<sub>2</sub> was associated with perimeter, Chl *a*, and emergent  
18 cover, but perimeter was not important when using a multi-  
19 variate statistical approach. Temporal variability in pCH<sub>4</sub> was  
20 linked with maximum depth and submerged plant cover, but  
21 maximum depth alone was the best-approximating model fol-  
22 lowing multivariate model selection.

23 Shallower systems had greater temporal variability in pCH<sub>4</sub>  
24 ( $R^2 = 0.18$ ,  $p = 0.01$ ,  $df = 27$ , Fig. 5D), again hinting at the role  
25 of mixing in driving variability. If a waterbody remains strati-  
26 fied or mixes daily throughout the summer, it is likely to dis-  
27 play relatively low variability in surface water dissolved CH<sub>4</sub>  
28 concentration. Those that mix intermittently (e.g., once a week  
29 or once per month) have longer time periods in which dis-  
30 solved oxygen can be depleted and CH<sub>4</sub> can build, and once  
31 mixing occurs, dissipation of this CH<sub>4</sub> will likely take several  
32 days, leading to fluctuating periods of high and low surface  
33 pCH<sub>4</sub>. Maximum depth plays an important role in regulating  
34 mixing as deeper waters mix less frequently (Holgerson  
35 et al. 2022). Vegetation may also contribute to greater stratifica-  
36 tion either by blocking wind (emergent vegetation) or through  
37 shading and dissipating kinetic wind energy (submerged vege-  
38 tation; Herb and Stefan 2004; Chimney et al. 2006; Andersen  
39 et al. 2017). In our mixed effects models, temporal variability  
40 in pCH<sub>4</sub> was negatively associated with greater submerged  
41 cover (though it did not meet criteria to be considered as an  
42 informative parameter). However, this negative relationship  
43 hints at vegetation's role in reducing mixing.

44 We only measured dissolved CH<sub>4</sub> in this study, which con-  
45 tributes to diffusive CH<sub>4</sub> emissions, and it is important to note  
46 that there may be similarly high temporal variability in  
47 ebullitive CH<sub>4</sub> emissions, which can contribute between 3%  
48 and 100% of the total CH<sub>4</sub> flux in waterbodies < 0.05 km<sup>2</sup>  
49 (estimated using data from Rosentreter et al. 2021b). The same  
50 factors that predict temporal variability of dissolved CH<sub>4</sub> are  
51 also likely to be important for diffusive CH<sub>4</sub>, with stratifica-  
52 tion and mixing controlling rates of production of CH<sub>4</sub> that  
53 can be released via ebullition, and plants possibly providing a  
54 physical block between ebullition and the atmosphere. The

methods we present here may be useful for identifying drivers 55  
of spatial and temporal variability in ebullitive CH<sub>4</sub> flux from 56  
small waterbodies. 57

### Implications for future upscaling of small waterbody CO<sub>2</sub> 58 and CH<sub>4</sub> emissions 59

60 Understanding drivers of spatial and temporal variability of 61  
pCO<sub>2</sub> and pCH<sub>4</sub> will inform better sampling strategies and 62  
help improve models that upscale greenhouse gas emissions 63  
from inland waterbodies. Here, we show that pCO<sub>2</sub> and pCH<sub>4</sub> 64  
within small waterbodies vary almost twice as much in time 65  
as in space. Furthermore, a single sample from a single loca- 66  
tion can misestimate mean seasonal pCO<sub>2</sub> and pCH<sub>4</sub> by up to 67  
44% for pCO<sub>2</sub> and up to 83.5% for pCH<sub>4</sub>. These misestimates 68  
demonstrate the importance of repeated sampling over time, 69  
followed by greater spatial coverage in small waterbodies. 70

71 There is still debate over the most appropriate sampling res- 72  
olution in space for accurate estimation of dissolved CO<sub>2</sub> and 73  
CH<sub>4</sub> concentrations and diffusive flux with the atmosphere. 74  
For example, recent work in tropical reservoirs in Brazil 75  
(Paranaíba et al. 2018), a hemiboreal lake in southern Sweden 76  
(Natchimuthu et al. 2017), and subarctic lakes in northern 77  
Sweden (Wik et al. 2016) recommend between 6 and 300 sam- 78  
pling locations per km<sup>2</sup>. Balancing a reasonable number of 79  
samples with accurately incorporating spatial variability is 80  
challenging. The low spatial variability of pCO<sub>2</sub> in ponds and 81  
shallow lakes recorded suggests a single sample can represent 82  
the entire waterbody on a given date. As pCH<sub>4</sub> was slightly 83  
more variable in space, more than one location in the 84  
waterbody should be sampled. While improving spatial resolu- 85  
tion of CO<sub>2</sub> and CH<sub>4</sub> dynamics in small waterbodies will 86  
improve upscaling estimates, this is of secondary importance 87  
to improved temporal resolution to improve pCO<sub>2</sub> and pCH<sub>4</sub> 88  
estimates from small lentic systems.

89 Sampling a waterbody repeatedly over time is necessary to 90  
accurately quantify seasonal patterns of dissolved CH<sub>4</sub> and 91  
CO<sub>2</sub> concentrations, though this is more important for CH<sub>4</sub> 92  
than CO<sub>2</sub>. Most measurements of dissolved gas concentrations 93  
and fluxes in temperate systems are made in the summer, and 94  
seasonal studies are often limited to a round of sampling in 95  
the spring, summer, and fall. This approach misses intra- 96  
seasonal variability, in addition to missing the transition 97  
period between seasons (i.e., the “shoulder seasons”) when 98  
important processes such as macrophyte die-off or spring thaw 99  
occur. For example, CH<sub>4</sub> emissions over a 2-week period in 100  
the late spring accounted for nearly 20% of annual CH<sub>4</sub> emis- 101  
sions from a 2.4 km<sup>2</sup> waterbody (Waldo et al. 2021), and CO<sub>2</sub> 102  
and CH<sub>4</sub> emissions during the ice-melt period represent 17% 103  
and 27% of annual emissions from northern lakes (Denfeld 104  
et al. 2018), highlighting the importance of short time periods 105  
between sampling events. Natchimuthu et al. (2017) suggest 106  
at least 8 sampling days during the ice-free season are needed 107  
to be within 20% of the true measure and Wik et al. (2016) 108  
suggest 11 sampling days. We recommend frequent sampling



1 particularly in smaller systems due to the relationship of  
2 increasing temporal variability of  $p\text{CH}_4$  and mean  $p\text{CH}_4$  as sys-  
3 tem size decreases.

4 Small and shallow waterbodies are known to release signifi-  
5 cant quantities of  $\text{CO}_2$  and  $\text{CH}_4$  to the atmosphere (Holgerson  
6 and Raymond 2016; Rosentreter et al. 2021a). Here, we have  
7 shown that the smallest of these systems also have the highest  
8 variability in  $p\text{CO}_2$  and  $p\text{CH}_4$  across space and time. Physical  
9 characteristics and dissolved nutrients appear to be the most  
10 important variables for understanding both mean  $p\text{CO}_2$  and  
11  $p\text{CH}_4$  and variability of  $p\text{CO}_2$  and  $p\text{CH}_4$  in space and time.  
12 Dissolved P concentration is particularly useful for under-  
13 standing  $\text{CO}_2$  dynamics—we found relationships between dis-  
14 solved P concentration and mean  $p\text{CO}_2$ , spatial variability in  
15  $p\text{CO}_2$ , and temporal variability in  $p\text{CO}_2$ . Physical features asso-  
16 ciated with regulation of mixing patterns, such as maximum  
17 depth, are important for predicting  $p\text{CH}_4$  and variability in  
18  $p\text{CH}_4$  and merit further investigation. Identifying variables to  
19 predict mean  $p\text{CH}_4$  and  $p\text{CO}_2$  and variability of  $p\text{CH}_4$  and  
20  $p\text{CO}_2$  over space and time in small waterbodies will inform  
21 future study designs and targeted sampling of variable sys-  
22 tems, and also reduce uncertainty in upscaling global green-  
23 house gas emissions.

#### 24 Data Availability Statement

25 The dataset used in this study can be accessed via the Figshare  
26 Repository ([https://figshare.com/articles/dataset/Dataset\\_for\\_Spatial\\_and\\_temporal\\_variability\\_in\\_greenhouse\\_gas\\_partial\\_pressures\\_in\\_shallow\\_lakes\\_and\\_ponds/19495121](https://figshare.com/articles/dataset/Dataset_for_Spatial_and_temporal_variability_in_greenhouse_gas_partial_pressures_in_shallow_lakes_and_ponds/19495121)) and the code  
27 used for statistical analysis is available on Github (<https://github.com/nray17/PONDING-GHG-R-Code>).  
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